## 84. Synthesis of Tricyclo [6.2.1.0<sup>1,5</sup>]undecadiones *via* Intramolecular Photoaddition of 5-(1-Cyclopentenylmethyl)-3-alkoxy-2-cyclopentenones

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## Summary

Irradiation of the dienone methyl ether **3a** furnished the unstable regioisomeric photoadducts **4a** and **5a** which on *Lewis* acid-catalyzed fragmentation afforded the tricyclic diketones **6**, **7** and **8**. The same mixture was obtained from the isobutyl. ether **3b** (*Scheme 2*).

**Introduction.** – The tricyclo [6.2.1.0<sup>1,5</sup>]undecane skeleton A (Scheme 1) is a basic structural feature of several natural products such as the zizaane type sesquiterpenes  $\mathbf{B}^{1}$ ) and the gibberellins<sup>2</sup>). The development of short and selective approaches to specificially functionalized ring systems of this type remains a fascinating challenge despite a number of previous successful syntheses<sup>1,2</sup>).



Continuing our studies of intramolecular *de Mayo* reactions  $[3]^3$ ) and their applications to the synthesis of natural products<sup>4</sup>) we envisaged the assembly of the system A by formation of the two C, C-bonds C(5), C(6) and C(1), C(10) in a single operation. Independently, *Pattenden et al.* have described a related construction of the bicyclo[3.2.1]octane system by subjecting 3-acetoxy-5-allyl-4-methyl-2-cyclopenten-1-ones to a photoaddition-retroaldol sequence [7]. However, we were unable to obtain 3-acyloxycyclopentenones **3c** regioselectively.

Preparation and irradiation of 5-(1-cyclopentenylmethyl)-3-methoxy-2-cyclopent-1-one (3a) (Scheme 2). – Kinetically controlled deprotonation [8] of the readily available enol ether 1a with lithium diisopropylamide followed by alkylation of the resulting enolate with the bromide 2 furnished cleanly the dienone 3a in 68%

<sup>&</sup>lt;sup>1</sup>) For literature regarding the isolation and structural elucidation of zizaane type sesquiterpenes and for previous synthetic work in this area see [1].

<sup>&</sup>lt;sup>2</sup>) For references concerning previous structural and synthetic work on gibberellins see [2].

<sup>3)</sup> For a review on intermolecular de Mayo reactions see [4].

<sup>&</sup>lt;sup>4</sup>) e.g. the syntheses of longifolene [5] and  $\beta$ -bulnesene [6].



yield. Irradiation of **3a** led to a mixture of unstable products from which the two major adducts **4a** and **5a** could be isolated. Their constitutions, tentatively assigned on the basis of IR., <sup>1</sup>H-NMR. and mass spectra, were ultimately confirmed by their fragmentation with boron trifluoride etherate<sup>5</sup>).

**Fragmentation of the photoproducts 4a and 5a.** – *Tamura et al.* have demonstrated the photoaddition of dimedone-O-butenyl ether and the cleavage of the resulting tricyclic  $\beta$ -alkoxyketone with a *Lewis* acid to give a bicyclic dihydro-furan [9]. Accordingly it might be expected that the photoproducts **4a** and **5a** would readily undergo a retro-aldol-like fragmentation under similar conditions. Indeed, treatment of **4a** with boron trifluoride etherate furnished smoothly the crystalline tricyclo [6.2.1.0<sup>1,5</sup>] undecadione **6**. Under identical conditions **5a** gave a mixture of the stereoisomers **7** and **8** which was separated by HPLC. For practical reasons it was preferable to treat the crude mixture of photoproducts directly with boron trifluoride etherate. Subsequent chromatography yielded the crystalline dione **6** in 24% and the isomers **7** and **8** in 31% overall yield from **3a**.

Structure of the products 6, 7 and 8. – On the basis of the <sup>1</sup>H-NMR. (360 MHz) and <sup>13</sup>C-NMR. (25.2 MHz) spectra the constitution of the diketones was readily assigned. In the <sup>1</sup>H-NMR, spectrum the least polar crystalline diketone 6 shows between  $\delta = 1.9$  and 2.6 ppm the signals of three methylene groups, each one coupled to one vicinal proton; in contrast, the diones 7 and 8 exhibit the signals

<sup>&</sup>lt;sup>5</sup>) The configurations of 4a and 5a were not investigated.

of one isolated methylene group. The configuration of **6** was not determined. However, the deshielding of  $H_A$  in **8** giving rise to a triplet at  $\delta = 2.63$  ppm indicates its *peri-syn*-planar position relative to the C(6)-carbonyl-O-atom. Since the dione **7** shows no signal downfield from  $\delta = 2.15$  ppm which couples with protons appearing upfield from  $\delta = 2.0$  ppm, the configurations of **7** and **8** were tentatively assigned as in *Scheme 2*. The <sup>13</sup>C-NMR. spectra are in agreement with these deductions. Thus, **6** shows a singlet at low field (58.6 ppm) corresponding to the quaternary C(1) (adjacent to a carbonyl group) whereas in the spectra of the diones **7** and **8** the C(1) singlet appears at 48.6 and 50.6 ppm, respectively. Further confirmation of these assignments follows from the interconversion of **7** and **8** by treatment of the isolated stereoisomers with boron trifluoride etherate.

**Discussion.** - The lack of regioselectively observed in the intramolecular photoaddition of 3a does not violate the 'rule of five'  $[10]^6$ ); both processes,  $3a \rightarrow 4a$ and  $3a \rightarrow 5a$  allow initial formation of a diradical intermediate possessing a five membered ring. In order to study the possible influence of steric effects the isobutyl enol ether 3b was subjected to the same photoaddition-fragmentation sequence giving a mixture of 6, 7 and 8 identical to that obtained from the methyl enol ether 3a. It thus appears that the comparable rate of C(2), C(1')- and C(3), C(1')bond formation is of electronic rather than of steric origin.

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## **Experimental Part**

General. All reactions were carried out under argon. The usual work-up procedure implies pouring the reaction mixture into a saturated aqueous solution of NaHCO<sub>3</sub>, extraction with ether or CH<sub>2</sub>Cl<sub>2</sub>, washing the combined organic layers with saturated aqueous NaCl solution, drying (Na<sub>2</sub>SO<sub>4</sub>) and removal of the solvent *in vacuo*. Thin layer chromatography (TLC.) was carried out on *Merck* 0.25 mm (60F254) silica gel plates. Preparative chromatography was carried out on silica gel (*Merck* 0.05–0.20 mm) unless otherwise specified. High pressure liquid chromatography (HPLC.) was performed on a *Waters* Associates apparatus M-6000 using  $\mu$ -Porasil ( $\mu$ 10, 4 mm ID×30 cm) columns and a R 401 refractometer detector. Gas chromatograms (GC.): 1 atm N<sub>2</sub>; glass columns (3 mm ID×3 m), stationary phases on chromosorb W (acid washed, 80/100 mesh): column 1: 5% SE 30; column 2: 5% OV 17; column 3: 5% OV 225; retention time in min (area %). Melting points (m.p.) are not corrected. IR. spectra: CCl<sub>4</sub> unless otherwise specified,  $\nu_{max}$  in cm<sup>-1</sup>. UV. spectra: methanol,  $\lambda_{max}$  in nm (loge). NMR. spectra: in CDCl<sub>3</sub>, internal standard tetramethylsilane ( $\delta$ =0 ppm); abbreviations: *s* singlet, *d* doublet, *t* triplet, *qa* quartet, *m* multiplet, br. broad, *J* spin-spin coupling constant (Hz), <sup>1</sup>H-NMR. at 100 MHz, unless otherwise specified, <sup>13</sup>C-NMR, at 25.2 MHz. Mass spectra (MS.): signals are given in *m/z* (rel. %).

5-(1-Cyclopentenylmethyl)-3-methoxy-2-cyclopenten-1-one (3a) (Scheme 2). – To a solution of dry diisopropylamine (0.45 g, 4.5 mmol) in dry THF (10 ml) were added with stirring first a 1.55 n solution of butyllithium in hexane (2.9 ml, 4.4 mmol) at  $-45^\circ$ , then after 30 min HMPA (0.82 g, 4.6 mmol), and finally, dropwise at  $-70^\circ$  a solution of 3-methoxy-2-cyclopenten-1-one<sup>7</sup>) (1a, 0.45 g, 4 mmol) in THF (3 ml). Stirring of the mixture for 15 min at  $-70^\circ$ , subsequent addition of 1-bromo-

<sup>&</sup>lt;sup>6</sup>) For a pertinent example see [9b].

<sup>7)</sup> The known enones 1a [11] and 1b [12] were obtained following a procedure for the preparation of 3-ethoxy-2-cyclopenten-1-one [13].

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methyl-1-cyclopentene (2) [14] (0.66 g, 4.1 mmol), further stirring at  $-70^{\circ}$  for 15 min, pouring of the reaction mixture into a saturated aqueous solution of NH<sub>4</sub>Cl followed by the usual work-up, chromatography (hexane/ethyl acetate 8:1) and distillation (140-145° (bath)/0.04 Torr) furnished the dienone 3a (oil, 0.52 g, 68%). – UV.: 240 (4.21). – IR. (film): 1699, 1600, 1360, 1290, 1245, 1170, 995. – <sup>1</sup>H-NMR.: 1.8–2.5 (8 H); 2.5–2.9 (3 H); 3.85 (s, 3 H); 5.30 (br. s, 1H); 5.40 (m, 1H). – MS.: 192 (43, C<sub>12</sub>H<sub>16</sub>O<sub>2</sub><sup>+</sup>), 111 (100), 112 (100).

Irradiation of 5-(1-cyclopentenylmethyl)-3-methoxy-2-cyclopentenone (3a). Irradiation of a solution of the dienone 3a (1.18 g, 6.15 mmol) in cyclohexane (150 ml) through Pyrex at 15° using a 125 W medium pressure Hg-lamp (*Philips*) for 7 h, followed by evaporation of the solution *in vacuo* gave a colourless oil (1.25 g). TLC. (toluene/ethyl acetate 7:3): 3 spots, Rf 0.65, 0.35 and 0.25. Chromatography of the unstable crude mixture (64 mg) on Al<sub>2</sub>O<sub>3</sub> (*Merck*, neutral, activity III, hexane/ether 8:2) furnished an unidentified compound (15 mg), followed by the more polar photoproduct 4a (oil, 20 mg). - IR.: 1745, 1290. - <sup>1</sup>H-NMR.: 1.2-2.4 (11 H); 2.55 (d, J=1.5, 1H); 2.66 (m, 1 H); 3.35 (s, 3 H). - MS.: 192 (26, C<sub>12</sub>H<sub>16</sub>O<sub>2</sub><sup>+</sup>), 178 (5), 164 (13), 152 (8), 149 (11), 112 (100), 111 (69), 109 (35).

Further elution afforded the most polar photoproduct 5a (oil, 7 mg). – IR.: 1760 br., 1080. – <sup>1</sup>H-NMR.: 1.2–2.6 (13 H); 3.15 (s, 3 H). – MS.: 192 (31, C<sub>12</sub>H<sub>16</sub>O<sub>2</sub><sup>+</sup>), 178 (59), 164 (18), 152 (18), 151 (16), 149 (48), 135 (57), 112 (100), 111 (80), 109 (61).

Tricyclo [6.2.1.0<sup>1,5</sup>]undecan-7, 10-dione (6) and tricyclo [6.2.1.0<sup>1,5</sup>]undecan-6,9-diones (7 and 8). a) Fragmentation of the crude photoproduct mixture. Boron trifluoride etherate (0.8 ml, 6.5 mmol) was added dropwise to a stirred solution of the crude photoproduct mixture (1.2 g) in dry benzene (200 ml) at  $+15^{\circ}$ . After stirring for 30 min the mixture was poured into a saturated aqueous solution of NH<sub>4</sub>Cl and subjected to the usual work-up. Distillation of a sample of the crude residue (108 mg) at 110-140° (bath)/0.01 Torr furnished a colourless oil (65 mg) which on GC. analysis (column 1, 150°) showed 4 major peaks: 11.0 (16.5), 12.1 (30), 14.2 (12), 17.0 (40); analogous results were obtained using the columns 2 and 3. HPLC. analysis showed 3 major peaks with 6.6, 9.0 and 10.2 min retention time in a ratio of ca. 2:1:1. Chromatography of the crude residue (hexane/ethyl acetate 2:1) gave the least polar diketone 6 (0.26 g, 24% yield from 3a), m.p. 54-56° (ether/pentane); GC. (column 1, 150°): 12.1. - IR.: 1743, 1725. - <sup>1</sup>H-NMR. (360 MHz): 1.35 (m, 1H); 1.52 (m, 1H); 1.7-1.9 (2 H); 1.95 (m, 1H); 2.0-2.15 (3 H); 2.23 (t, J=15, 1H); 2.30 ( $d \times d$ , J=4 and 19, 1H; irradiation at 2.05  $\rightarrow d$ , J=19; 2.35 ( $d \times d$ , J=5 and 13, 1H; irradiation at  $3.02 \rightarrow d$ , J=13); 2.47 ( $d \times d$ , J=7 and 19, 1H; irradiation at  $3.02 \rightarrow d$ , J = 19; 2.60 ( $d \times d$ , J = 5 and 15, 1H; irradiation at  $2.05 \rightarrow d$ , J = 15); 3.02  $(d \times d, J=5 \text{ and } 7, 1\text{H}; \text{ irradiation at } 2.35 \rightarrow d, J=7)$ . - <sup>13</sup>C-NMR.: 216.3s, 209.9s, 58.6s, 49.5d, 46.8d, 42.4t, 41.5t, 39.7t, 29.5t, 27.4t, 23.9t. - MS.: 178 (90,  $C_{11}H_{14}O_2^+$ ), 160 (20), 135 (79), 95 (31), 81 (100), 79 (42).

Further elution gave fractions containing the diketones 6, 7 and 8 (50 mg) followed by a fraction containing the stereoisomers 7 and 8 (0.34 g, 31% yield from 3a; GC., column 1, 150°: 16.4) and, finally, unchanged dienone 3a (50 mg). Part of the fractions containing the isomers 7 and 8 (100 mg) was separated by preparative HPLC. using 2  $\mu$ -Porasil columns in series (hexane/ethyl acetate 6:1, flow rate 4.0 ml, min<sup>-1</sup>, 2400 psi) to give the less polar isomer 7 (oil), retention time 9.2 min. GC. (column 1, 150°): 16.2. – IR.: 1753, 1713. – <sup>1</sup>H-NMR. (360 MHz): 1.7–2.05 (7 H); 2.12 (m, 1H); 2.34 (d, J = 18, 1H; irradiation at 2.46  $\rightarrow$  br. s); 2.46 (m, 1H); 2.48 (d, J = 18, 1H); 2.55–2.70 (2 H); 2.71 (m, 1H). – <sup>13</sup>C-NMR.: 218.8s, 210.4s, 61.0d, 51.3t, 48.6s, 46.2d, 44.8t, 38.1t, 36.3t, 29.1t, 23.6t. – MS.: 178 (97, C<sub>11</sub>H<sub>14</sub>O<sub>2</sub>+), 135 (27), 81 (100).

Further elution (HPLC.) afforded the more polar isomer **8** (oil), retention time 10.3 min. GC. (column 1, 150°): 17.05. – IR.: 1757, 1724. – <sup>1</sup>H-NMR. (360 MHz): 1.55–1.95 (6 H); 1.90 (br. d, J = 19, 1 H; irradiation at 2.17  $\rightarrow$  br.s); 2.07 ( $d \times d$ , J = 4 and 12, 1 H; irradiation at 2.34  $\rightarrow d$ , J = 4); 2.17 ( $d \times d$ , J = 4 and 19, 1 H; irradiation at 1.90  $\rightarrow t$ , J = 4); 2.34 ( $d \times d \times d$ , J = 2 and 5 and 12, 1 H; irradiation at 2.07  $\rightarrow m$ ; irradiation at 2.83  $\rightarrow d \times d$ , J = 2 and 12); 2.42 ( $d \times d \times d$ , J = 2, 4 and 15, 1 H; irradiation at 2.83  $\rightarrow d \times d$ , J = 2 and 15); 2.54 ( $d \times d$ , J = 4 and 15, 1 H; irradiation at 2.83  $\rightarrow d \times d$ , J = 2 and 15); 2.63 (t, J = 9, 1 H; irradiation at 1.83  $\rightarrow$  br. s); 2.83 (qa, J = 4, 1 H). – <sup>13</sup>C-NMR.<sup>8</sup>): 214.9s, 207.1s, 61.4d, 50.6s, 50.1d, 45.5t, 44.8t, 40.3t, 36.4t, 21.4t, 20.8t. – MS.: 178 (100, C<sub>11</sub>H<sub>14</sub>O<sub>2</sub><sup>+</sup>), 135 (90), 108 (53), 93 (35), 81 (80), 79 (48), 67 (28), 55 (29).

<sup>&</sup>lt;sup>8</sup>) The <sup>13</sup>C-NMR. signals of 8 were assigned by subtracting the spectrum of 7 from that of a mixture of 7 and 8.

b) Fragmentation of the isolated photoproducts. Boron trifluoride etherate (1 drop) was added to a solution of the isolated photoproduct 4a (7 mg) in dry benzene (3 ml). The mixture was stirred at 20° for 20 min to give, after the usual work-up and bulb-to-bulb distillation, the diketone 6, identified by TLC. (toluene/ethyl acetate 7:3) and GC. (column 1). An identical experiment with 5a furnished a mixture of the diketones 7 and 8.

Acid-catalyzed interconversion of the diketones 7 and 8. - a) By silica gel. A solution of 7 (4.6 mg, isolated by HPLC.) was stirred with silica gel (0.2 g) in ethyl acetate (3 ml) at 20° for 15 h. Filtration and evaporation of the solution gave an oil (4.4 mg) which was unchanged 7 (HPLC.: 2  $\mu$ -Porasil columns in series, hexane/ethyl acetate 4:1, flow rate 4 ml, min<sup>-1</sup>, 2200 psi). An identical experiment with 8 (4.2 mg) gave a 1:9 mixture (4.2 mg, analyzed by HPLC. as above) of 7 and 8.

b) By boron trifluoride etherate. Boron trifluoride etherate (2 drops) was added to a solution of 7 (4.4 mg) in dry benzene (0.5 ml). The solution was stirred at 20° for 30 min to give after the usual work-up a 9:1-mixture (4.2 mg, analyzed by HPLC. as described above) of 7 and 8. An identical experiment with 8 (4.2 mg) gave a 1:4 mixture (4.0 mg, analyzed by HPLC. as above) of 7 and 8.

Preparation of the isobutoxydienone 3b and its conversion to the diketones 7 and 8. – 5-(1-Cyclopentenylmethyl)-3-isobutoxy-2-cyclopenten-1-one (3b). 3-lsobutoxy-2-cyclopenten-1-one<sup>7</sup>) 1b (0.30 g, 2.0 mmol) was successively treated with lithium diisopropylamide and 1-bromomethyl-1-cyclopentene (2) as described above for the transformation  $1a \rightarrow 3a$  to give after bulb-to-bulb distillation (bath 140–150°/0.22 Torr) the dienone 3b as a colourless oil (0.31 g, 67%). – IR. (film): 1700, 1600, 1355, 1175, 1000. – <sup>1</sup>H-NMR.: 1.00 (d, J = 7, 1H); 1.75–2.45 (9 H); 2.5–2.85 (3 H); 3.75 (d, J = 6, 2 H); 5.25 (s, 1 H); 5.40 (br. s, 1 H).

b) Conversion of the dienone **3b** into the diketones **6**, 7 and **8**. Irradiation of a solution of **3b** (191 mg, 0.81 mmol) in cyclohexane (150 ml) through Pyrex at 15° using a 125 W medium pressure Hg-lamp (*Philips*) for 9 h followed by evaporation of the solution gave a colourless oil (196 mg), TLC. (toluene/ethyl acetate 9:1): 3 spots, Rf 0.75, 0.50 and 0.45. Boron trifluoride etherate (2 drops) was added to stirred solution of the crude mixture (136 mg) in dry benzene (50 ml) at  $+10^{\circ}$ . The solution was stirred for 30 min then poured into a saturated aqueous solution of NH<sub>4</sub>Cl and subjected to the usual work-up to give an oil (107 mg). GC. (column 1, 140°) of this oil showed 4 peaks: 15.2 (19), 16.7 (33), 20.8 (9) and 24.9 (38) which correspond to the crude products obtained from **3a**. Chromatography (hexane/ethyl acetate 9:1) of the crude mixture (100 mg) gave the dione **6** (14 mg), fractions containing **6**, 7 and **8**, followed by a mixture of 7 and **8** (39 mg) identified by TLC. (hexane/ethyl acetate 3:7) and GC. (column 1) comparison.

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